

Supporting Material

Thermodynamic Studies of $[\text{H}_2\text{Rh}(\text{diphosphine})_2]^+$ and $[\text{HRh}(\text{diphosphine})_2(\text{CH}_3\text{CN})]^{2+}$

Complexes in Acetonitrile

Aaron D. Wilson, Alexander J. M. Miller, Daniel L. DuBois[‡], Jay A. Labinger, and John E.

Bercaw

Arnold and Mabel Beckman Laboratories of Chemical Synthesis, California Institute of Technology, Pasadena, California 91125

[‡]Chemical and Materials Sciences Division, Pacific Northwest National Laboratory, Richland, WA 99352

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Derivation of Tolman cone angles. Tolman calculated the cone angle for diphosphine ligands (Θ_B) using equation 1, where Θ_M is the cone angle of the corresponding monodentate PR_3 ligand with the same R substituents as the diphosphine ligands and α is the experimentally observed PMP bond angle of the chelating diphosphine ligand. We have extended this method to the other ligands shown in Chart 1 using the experimentally determined bite angles from the work of Aguila, et. al. and the cone angles for the corresponding monodentate ligands used by Tolman.² The cone angle values derived in this way agree very well with the available data from Tolman (values shown in parentheses) and allow a logical extension to the new ligands used in this work.

$$\Theta_B = 2/3(\Theta_M + 1/2\alpha) \quad (1)$$

Table S1. Data used to calculate Tolman cone angles for bidentate ligands and $\Sigma^3\chi$.

PP	$1/2 \alpha (^{\circ})^a$	$\Theta_M (^{\circ})^b$	$\Theta_B (^{\circ})^c$	TEM ^d $\Sigma^3\chi$
dppm	36	145	121 (121)	11.2
dmpe	43	118	107 (107)	7.8
depe	43	132	117 (115)	6.2
dppe	43	145	125 (125)	11.2
dcpe	43	170	142 (142)	2.8
depp	46	132	119	6.2
depx	50	132	121	6.2
dppb	43	145	125	12.9
dppp	46	145	127 (127)	11.2

^a Data from Ref 1. ^b Data from reference 2. ^c Calculated using eq. 1. Values in parenthesis are from ref. 2.

^d The Tolman Electronic Parameters (TEMs, χ) values for substituents were obtained from Ref. 2. The sum (Σ^3) of χ for each phosphine was obtained by adding the values of the two substituents to the values for CH₃ or Ph. The χ values for CH₃ and Ph were used for the alkyl or aryl groups of the diphosphine backbone.

References for Supporting Information

1. Aguila, D.; Escribano, E.; Speed, S.; Talancon, D.; Yerman, L.; Alvarez, S. *Dalton Trans.* **2009**, 6610-6625.
2. Tolman, C. A. *Chem. Rev.* **1977**, 77, 313-348.